

# Where To Download Stoichiometry Gizmo Quiz Answers

## Stoichiometry Gizmo Quiz Answers

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Gizmo: Stoichiometry Stoichiometry Gizmo Quiz Answers.pdf stoichiometry gizmo worksheet answers stoichiometry gizmo worksheet answerstake part in a chemical reaction. to begin, check that this equation is shown:  $\text{Fe}_2\text{O}_3 + 3\text{CO} \rightarrow 2\text{Fe} + 3\text{CO}_2$  1. look at the coefficients (such as the Page 11/30

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The Stoichiometry Gizmo™ allows you to try your hand at figuring out the amounts of reactants and products that take part in a chemical reaction. To begin, check that



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Answers  
this equation is shown:  $Fe_2O_3 + 3CO \rightarrow 2Fe + 3CO_2$ . Look at the coefficients (such as the “3” in  $3CO$ ) in front of each substance in the equation.

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Answers  
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Since 20 moles NO<sub>2</sub> are produced, and it takes 3 moles O<sub>2</sub> to produce 4 moles of NO<sub>2</sub>, the answer would be 15 moles.  $15 \text{ moles} = 20 \text{ moles NO}_2 \times (3 \text{ moles O}_2 / 4 \text{ moles NO}_2)$ --- This is the molar ratio between the two.

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## Answers

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