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Objectives. After completing this section,
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equilibrium constant expression for a given reaction. assess, qualitatively, how far a reaction will proceed in a given direction, given the value of K_{eq} ; explain the difference between rate and equilibrium.

6.8: Describing a Reaction - Equilibria,

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latter fitted better. The sorption-desorption of PFOA onto ten soil samples depended on soil organic carbon content and composition of soil minerals.

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developed as a correlation of reactivities

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(rates) and equilibria in reactions of meta- and para-substituted benzene derivatives.

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$A + \nu_B B \rightleftharpoons \nu_R R + \nu_S S$ [2.1] . . . the

system inevitably moves toward a state of

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