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Reality - Limiting Reactant 1. Write the balanced equation

for the reaction between hydrogen and oxygen. Balanced

Equation: $2\text{H}_2 + 1\text{O}_2 \rightarrow 2\text{H}_2\text{O}(\text{g})$ Suppose that 4 molecules of

hydrogen gas and 4 molecules of oxygen gas react to form

water. Make a drawing that represents the reaction container

before and after the reaction. $\text{O O O O O O O O O O O O}$

After Before How many molecules of water can be produced?

4 molecules Which reactant is in excess?

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Worksheet 2 - Measuring Pressure . Problem 1 and 2
Calculate the pressure using the barometer . 1. 720 mmHg =
0.95 atm. 2. 760 mmHg = 1 atm. Problems 3 and 4.
Calculate the pressure of the gas in the flask connected to
the manometer. 3. a. $127 - 84 = 43 \text{ mmHg}$ $730 + 43 = 773$
mmHg

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List at least 3 Physical Properties of Ionic Compounds: 2.
Why must the total amount of positive charge in an ionic
compound equal the ...

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and 9 (central force particle model, energy transfer model,
and impulsive force model): archive of teacher contributions
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pair of compounds to help you suggest formulas that
account for these ratios. 1. Compounds of carbon and~~

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oxygen Compound A: 57.1 g O / 42.9 g C Compound B: 72.7 g O and 27.3 C a. Determine the value of the ratio mass O mass C in each ...

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Worksheet 3 Ionic Compounds Properties Ionic compounds generally have higher mp ' s and bp ' s than do molecular compounds. When molten,

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building (using the model kit included in the Lab Kit) a variety of simple molecules and making observations on their structure. Your task will be to develop a better understanding of molecules and their 3-D properties.
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Unit 10: Solutions Unit 11: Acids & Bases. Acids, Bases & pH.

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Name Date Pd Unit 6 – Worksheet 3 Ionic Compounds
Properties Ionic compounds generally have higher mp ' s
and bp ' s than do molecular compounds.

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