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Question 1 Syllabus References: A:

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potential and gave uses of the values of standard potential.

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is solution containing 6.22 g of an acid  
 $H_2Y$  per  $dm^3$ . B contains 3.90 g of  
NaOH per  $dm^3$  of solution. (a) Put A  
into the burette and titrate it against



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20.0 cm<sup>3</sup> or 25.0 cm<sup>3</sup> portions of B using methyl orange as indicator. Repeat the titration to obtain consistent titres. Tabulate your

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