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$-1R = 8.31 \text{ J K mol}^{-1}$ (from IB Data booklet) Use $PV = nRT$ (from IB Data booklet) $V = nRT/P = 1.50 \times 8.31 \times 298.15 / 1.08 \times 1.01 \times 10^5 = 30.034.1 \text{ m}^3 = 34.1 \text{ dm}^3$ 3[4] Award [4] for correct final answer. Award [3] (max) for 0.0341 dm^3 or 22.7 dm^3 . 3Award [3] (max) for 34.4 dm^3 . 3Award [2] (max) for 22.9 dm^3 . 3Award [2] (max) for 0.0227 dm^3 .

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